

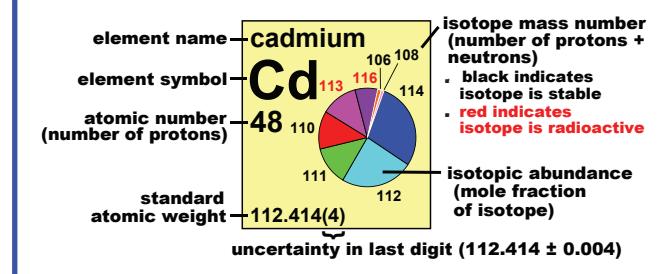
IUPAC Periodic Table of the Isotopes

1 hydrogen H 1 [1.007 84, 1.008 11]	2 lithium Li 3 [6.938, 6.997]	2 beryllium Be 4 9.012 1831(5)
sodium Na 11 22.989 769 28(2)	magnesium Mg 12 [24.304, 24.307]	
potassium K 19 39.0983(1)	calcium Ca 20 40.078(4)	scandium Sc 21 44.955 908(5)
rubidium Rb 37 85.4678(3)	strontium Sr 38 87.62(1)	yttrium Y 39 88.905 84(2)
caesium (cesium) Cs 55 132.905 451 96(6)	barium Ba 56 137.327(7)	lanthanoids 57 - 71 178.49(2) 178
francium Fr 87 88		rutherfordium Rf 104 actinoids 89 - 103 104
dubnium Db 105 106	seaborgium Sg 107 108	bohrium Bh 109 110
	hafnium Hf 72 178.947 88(2)	tantalum Ta 73 180.947 88(2)
	tungsten W 74 183.84(1)	rhenium Re 75 186.207(1)
	osmium Os 76 190.23(3)	iridium Ir 77 192.217(3)
	platinum Pt 78 195.084(9)	gold Au 79 196.966 569(5)
	mercury Hg 80 200.592(3)	mercury Hg 80 204.382, 204.385
	thallium Tl 81 204.382, 204.385	thallium Tl 81 207.2(1)
	lead Pb 82 208.980 40(1)	lead Pb 82 207.2(1)
	bismuth Bi 83 208.980 40(1)	bismuth Bi 83 208.980 40(1)
	polonium Po 84 85	polonium Po 84 85
	astatine At 85 86	astatine At 85 86
	radon Rn 86 87	radon Rn 86 87

Element Background Color Key

Standard atomic weights are the best estimates by IUPAC of atomic weights that are found in normal materials, which are terrestrial materials that are reasonably possible sources for elements and their compounds in commerce, industry, or science. They are determined using all stable isotopes and selected radioactive isotopes (having relatively long half-lives and characteristic terrestrial isotopic compositions). Isotopes are considered stable (non-radioactive) if evidence for radioactive decay has not been detected experimentally.

- Element has two or more isotopes that are used to determine its standard atomic weight. The isotopic abundances and atomic weights vary in normal materials. These variations are well known, and the standard atomic weight is given as lower and upper bounds within square brackets, [].
- Element has two or more isotopes that are used to determine its standard atomic weight. The isotopic abundances and atomic weights vary in normal materials, but upper and lower bounds of the standard atomic weight have not been assigned by IUPAC or the variations may be too small to affect the standard atomic weight value significantly. Thus, the standard atomic weight is given as a single value with an uncertainty that includes both measurement uncertainty and uncertainty due to isotopic abundance variations.
- Element has only one isotope that is used to determine its standard atomic weight. Thus, the standard atomic weight is invariant and is given as a single value with an IUPAC evaluated uncertainty.
- Element has no standard atomic weight because all of its isotopes are radioactive and, in normal materials, no isotope occurs with a characteristic isotopic abundance from which a standard atomic weight can be determined.



13 boron B 5 [10.806, 10.821]	14 carbon C 6 [12.0096, 12.0116]	15 nitrogen N 7 [14.00643, 14.00728]	16 oxygen O 8 [15.99903, 15.99977]	17 fluorine F 9 18.998 403 163(6)	18 helium He 2 4.002 602(2)
aluminium (aluminum) Al 13 26.981 5385(7)	silicon Si 14 [28.084, 28.086]	phosphorus P 15 30.973 761 998(5)	sulfur S 16 [32.059, 32.076]	chlorine Cl 17 [35.446, 35.457]	neon Ne 10 20.1797(6)
gallium Ga 31 69.723(1)	germanium Ge 32 72.630(8)	arsenic As 33 74.921 595(6)	selenium Se 34 78.971(8)	bromine Br 35 [79.901, 79.907]	krypton Kr 36 83.798(2)
indium In 49 114.818(1)	tin Sn 50 118.710(7)	antimony Sb 51 121.760(1)	tellurium Te 52 127.60(3)	iodine I 53 126.904 47(3)	xenon Xe 54 131.293(6)
gadolinium Gd 64 157.25(3)	europium Eu 63 151.964(1)	thallium Tl 81 150.36(2)	thallium Tl 81 151.25(3)	lead Pb 82 158.925 35(2)	radon Rn 86 164.930 33(2)
terbium Tb 65 158.925 35(2)	cerium Ce 60 140.907 66(2)	curium Cm 95 151.964(1)	curium Cm 95 152.25(3)	berkelium Bk 97 158.500(1)	cerium Ce 60 158.500(1)
dysprosium Dy 66 162.500(1)	praseodymium Pr 59 140.907 66(2)	americium Am 95 151.964(1)	americium Am 95 152.25(3)	einsteinium Es 99 164.930 33(2)	cerium Ce 60 164.930 33(2)
holmium Ho 67 164.930 33(2)	neodymium Nd 60 144.242(3)	curium Cm 96 151.964(1)	curium Cm 96 152.25(3)	fermium Fm 100 167.259(3)	cerium Ce 60 167.259(3)
erbium Er 68 167.259(3)	lanthanum La 57 138.905 47(7)	berkelium Bk 97 158.925 35(2)	berkelium Bk 97 159.25(3)	mendelevium Md 101 167.259(3)	cerium Ce 60 167.259(3)
thulium Tm 69 168.934 22(2)	cerium Ce 58 140.116(1)	californium Cf 98 164.930 33(2)	californium Cf 98 165.25(3)	nobelium No 102 167.259(3)	cerium Ce 60 167.259(3)
ytterbium Yb 70 173.054(5)	cerium Ce 58 140.116(1)	einsteinium Es 99 164.930 33(2)	einsteinium Es 99 165.25(3)	lawrencium Lr 103 174.9668(1)	cerium Ce 60 167.259(3)
lutetium Lu 71 174.9668(1)					

lanthanum La 57 138.905 47(7)	cerium Ce 58 140.116(1)	praseodymium Pr 59 140.907 66(2)	neodymium Nd 60 144.242(3)	promethium Pm 61 144.242(3)	samarium Sm 62 150.36(2)	europeum Eu 63 151.964(1)	gadolinium Gd 64 151.964(1)	terbium Tb 65 151.964(1)	dysprosium Dy 66 151.964(1)	holmium Ho 67 151.964(1)	erbium Er 68 151.964(1)	thulium Tm 69 151.964(1)	ytterbium Yb 70 151.964(1)	lutetium Lu 71 151.964(1)
actinium Ac 89 89	thorium Th 90 232.0377(4)	protactinium Pa 91 231.035 88(2)	uranium U 92 238.028 91(3)	neptunium Np 93 238.028 91(3)	plutonium Pu 94 239.028 91(3)	americium Am 95 240.028 91(3)	curium Cm 96 242.028 91(3)	berkelium Bk 97 244.028 91(3)	californium Cf 98 247.028 91(3)	einsteinium Es 99 247.028 91(3)	fermium Fm 100 247.028 91(3)	mendelevium Md 101 247.028 91(3)	nobelium No 102 247.028 91(3)	lawrencium Lr 103 247.028 91(3)